Study programme

Part A) of the study programme *

Learning outcomes

Faculty offering the field of study:		Faculty of Chemistry
Field of study:		Chemistry
Level of study:		First-cycle studies
Level of the Polish Qualification Framework:		Level 6
Profile:		General academic
Professional degree awarded to the graduate:		Licencjat
Allocation of the field of study within academic or artistic discipline(s), to which learning outcomes for a given field of study		Disciplines: Chemical Sciences (100 %)
refer::		Major discipline: Chemical Sciences
Symbol	Upon completion the graduate achieves the	e learning outcomes specified below:
KNOWLEDGE		
K_W01	The graduate has advanced knowledge of principles	of chemistry and the nomenclature.
K_W02	The graduate knows most important chemical eleknows ways of correlations between elements compounds, and the place of the element in the periods.	ements and their compounds. The graduate of properties and their primary chemical
K_W03	The graduate has advanced knowledge of the principles of linear algebra, mathematical analysis and statistics necessary for the description and modelling of chemical phenomena.	
K_W04	The graduate knows the role of experiments and cor	nputer simulations in chemical processes.
K_W05	The graduate knows basic software packages for the	
K_W06	The graduate knows theoretical and practical aspectanalysis by means of conventional and instrumen rules.	
K_W07	The graduate has advanced knowledge of functional mechanisms.	al groups of organic compounds and reaction
K_W08	The graduate knows states of matter, equation intermolecular forces, laws of thermodynamic electrochemistry.	
K_W09	The graduate is familiar with basic concepts, terms, laws and rules of physics and with their universal character to the extent sufficient to continue education.	
K_W10	The graduate knows basic concepts and research methods of contemporary inorganic and coordination chemistry.	
K_W11	The graduate knows the basics of biochemistry and chemism of metabolic processes.	
K_W12	The graduate knows techniques of collecting and preparing samples from environmental matrices for the analysis, water quality indicators, toxicity texts and ways of neutralisation of sewage.	
K_W13	The graduate has advanced knowledge of aspects of the construction and methods of assessment of properties of chemical materials and substances. The graduate has knowledge enabling them to use materials for a specified practical purpose and to indicate a method of their management after their lifetime.	

K_W14	The graduate knows and understands theoretical fundamentals of quantum chemistry, postulates
_	of quantum mechanics and their application to the description of atoms and molecules. The
	graduate knows and understands theoretical fundamentals of various molecular spectroscopies.
K_W15	The graduate has knowledge of chemical technology and engineering.
K_W16	The graduate is aware of occupational health and safety regulations and basic concepts in
	toxicology. The graduate knows legal regulations pertaining to standards and requirements
	binding in chemical laboratories as well as legal regulations concerning hazardous substances
	their storage and labelling.
	SKILLS
K U01	The graduate is able to use the nomenclature of chemistry and concepts in general chemistry.
K_U02	The graduate is able to correlate properties of chemical elements and their chemical compounds
K_002	with their place in the periodic table and to correlate chemical properties of substances with their
	contemporary applications.
K_U03	The graduate is able to apply the methods of linear algebra and mathematical analysis in selected
	issues in physics and chemistry.
K_U04	The graduate displays the ability to describe and model chemical phenomena and uses selected
11_00.	numerical processes in chemical computations.
K_U05	The graduate is able to perform basic measurements of chemical quantities and is able to
	elaborate results of physicochemical experiments.
K_U06	The graduate is able to perform quantitative analyses using gravimetric, volumetric and
K_000	instrumental methods on the basis of analytical procedures. The graduate is able to prepare ar
	analysis-based report.
K_U07	The graduate is able to recognise functional groups of organic compounds and to perform
	experiments in organic chemistry.
K_U08	The graduate recognises states of matter and is able to define and describe physicochemica
	processes.
K_U09	The graduate is able to develop simple physical experiments, analyse their results and explain
	physical phenomena occurring in the surrounding world. The graduate is able to solve basic
	problems relying on the laws of physics.
K_U10	The graduate is able to synthesise and decompose simple inorganic compounds and selected
	coordination compounds.
K_U11	The graduate is able to describe the structure and functions of macromolecular compound
	occurring in living organisms and to specify metabolic transformations occurring in major
	metabolic pathways as well as ways of storing and processing chemical energy in the cell.
K_U12	The graduate is able to collect and prepare environmental samples and to analyse them.
K_U13	The graduate is able to find correlations between material behaviour during formation and use
	and physicochemical properties, structure and structural type.
K_U14	The graduate is able to use basic quantum numerical methods to describe, in qualitative terms
17 1117	properties, structures and reactivity of chemical systems.
K_U15	The graduate is able to solve basic problems related to the completion of technological
IZ 111.6	processes.
K_U16	The graduate is able to behave properly while facing a variety of emergences, such as fire o
V 1117	contact with chemical reagents.
K_U17	The graduate displays language skills in a modern foreign language at the intermediate level (B2
	level) in daily life, in education-related situations and while preparing their diploma paper.
	SOCIAL COMPETENCES
K_K01	Analytical thinking: The graduate is able to work on his/her own and effectively with large
_	amounts of data, to perceive interrelations between phenomena and draw correct conclusions
	using the principles of logic.
K_K02	Creativity: The graduate thinks creatively in order to improve existing solutions or develop new
_	ones.
	Conscientiousness and accuracy: The graduate strives to complete a task as effectively a
K_K03	conscientiousness and accuracy. The graduate strives to complete a task as effectively as

K_K04	Communication skills: The graduate is able to communicate the achievements of chemical	
	knowledge to other persons effectively and clearly. The graduate adjusts the level and form of	
	presentations to the needs and capabilities of receivers.	
K_K05	Pursuit of development: The graduate is focused on the constant acquisition of new knowled	
	skills and experience. The graduate acknowledges the need for constant self-improvement and	
	increasing his/her professional skills. The graduate is aware of the limitations of their knowledge	
	and understands the need for further education.	
K_K06	Perseverance and consistency: The graduate works systematically and has a positive attitude to	
	obstacles standing in the way of reaching the desired objectives. The graduate observes	
	deadlines. The graduate understands the need to be systematic in all projects undertaken.	
K K07	Autonomy: The graduate implements agreed objectives on his/her own, taking autonomous and	
_	sometimes difficult decisions. The graduate is able to find information in the field literature.	
K_K08	Professionalism and ethics: The graduate knows and abides by the standards binding for	
	chemists, including ethical standards. The graduate understands the social role of the profession.	
	The graduate understands and recognises the importance of intellectual honesty and integrity,	
	care of one's health and of the natural environment in activities undertaken by themselves and by	
	other persons.	
K_K09	Team work: The graduate is able to establish and maintain long-term and effective collaboration	
_	with other persons. The graduate endeavours to achieve the objectives of the team by proper	
	planning and organisation of their own work and the work of other persons. The graduate	
	motivates collaborators to increase their efforts in order to achieve the assumed objectives.	